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The formation of H₂: experiments on an amorphous silicate surface



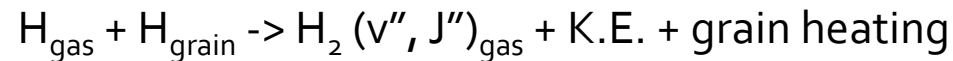
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Index

- Brief timeline of H₂ formation experiments.
- The experimental setup
 - The atomic beam
 - The sample
 - The detectors
- Internal excitation of nascent H₂
- Experiments (I): T_{gas} on H₂ formation
- Experiments (II): H₂ pre-coverage on H₂ formation
- Conclusions

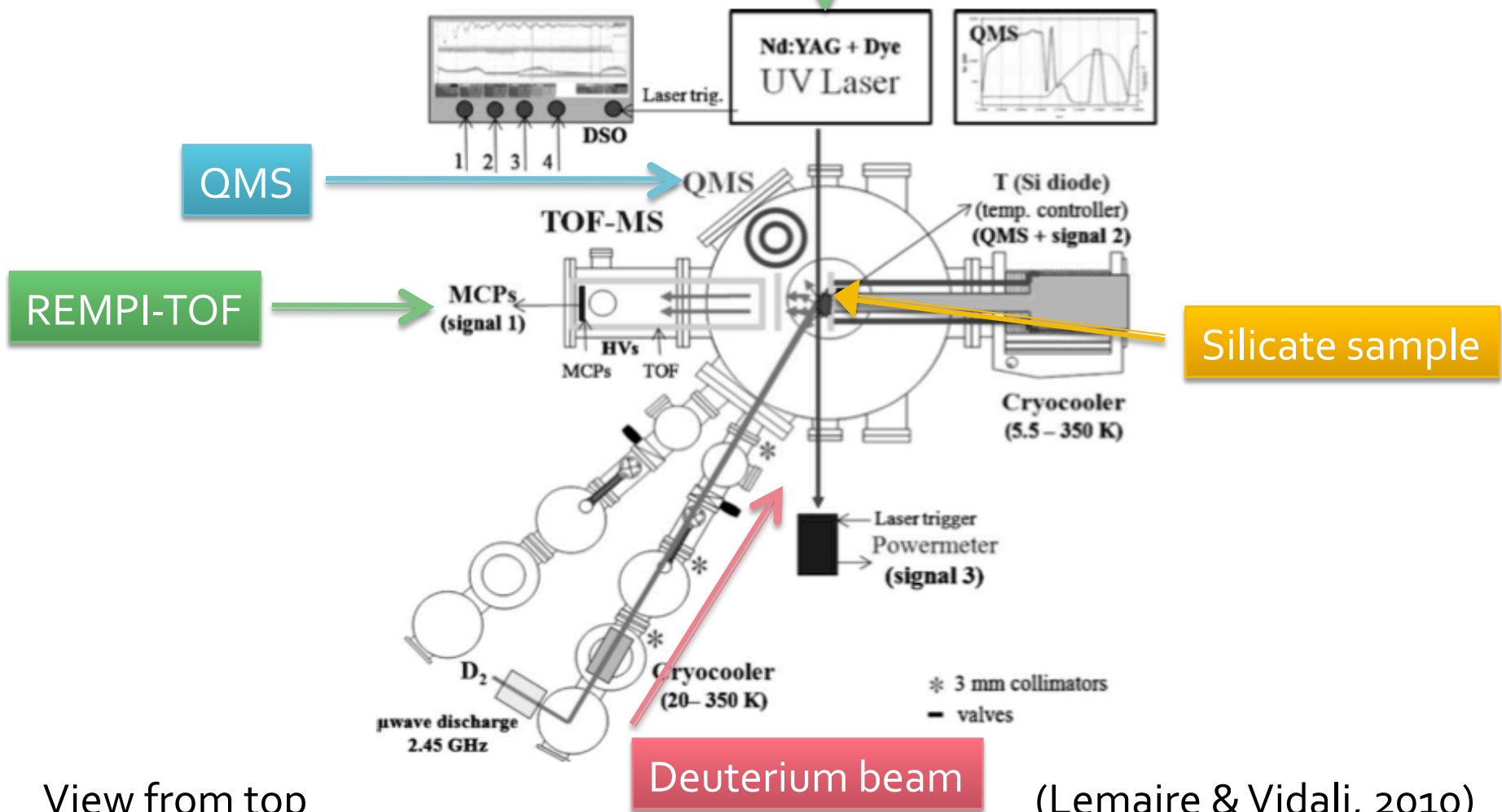
A brief timeline: experiments



- HD on polycrystalline silicates (Pirronello et al. 1997): efficient up to $T_{\text{dust}} < 10 \text{ K}$.
- HD on amorphous carbon (Pirronello et al. 1999).
- HD on amorphous water ice (Manico et al. 2001).
- HD on amorphous ice at 8 K (Hornekaer et al. 2003).
- HD on amorphous silicate (Perets, 2007): efficient up to $T_{\text{dust}} < 14 \text{ K}$.
- HD formed on graphite held at 15 K (Latimer et al 2008).
- D_2 formed on silicate held at 10 K efficient to $T_{\text{dust}} < 70 \text{ K}$. (Lemaire & Vidalı 2010).

The experimental setup

UV Laser (220 nm)

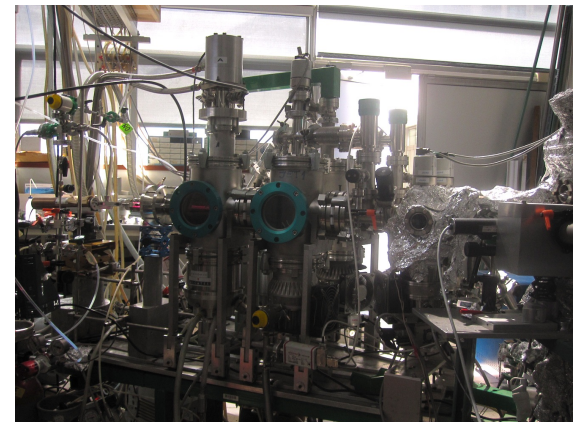
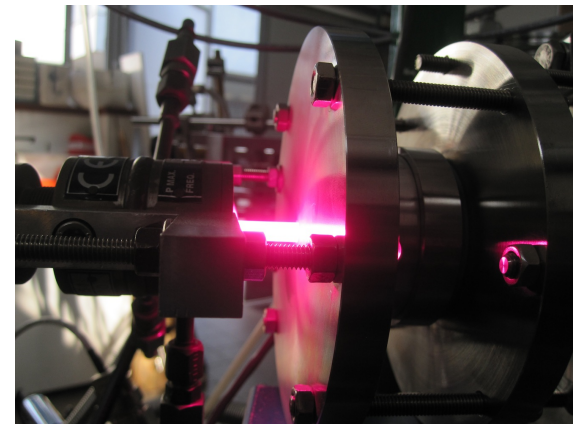


View from top

(Lemaire & Vidali, 2010)

The atomic beam

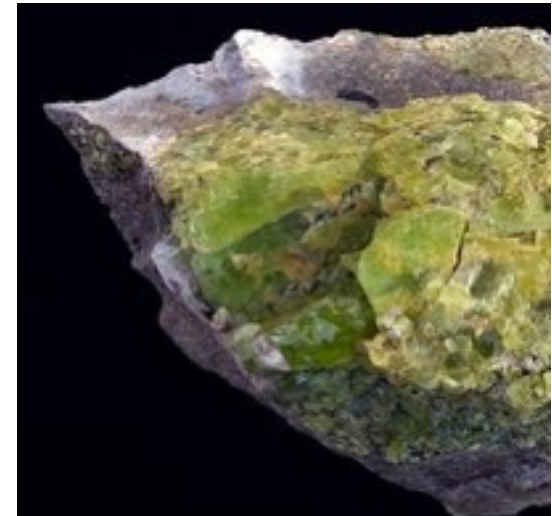
- We start with D_2 .
- A microwave discharge gives a dissociation rate up to 70%.
- D atoms are sent towards our sample via a triple differentially pumped beam and reach the UHV chamber at 10^{-10} mbar.
- The surface is seen at an angle of $\sim 41^\circ$ to the beam, allowing detection of $41/180$ of the molecules and a density of $\sim 2.9 \times 10^7$ molecules cm^{-3} in all states.
- $T_{\text{beam}} = 20 - 300 \text{ K}$



FORMOLISM @ Cergy-Pontoise

The sample

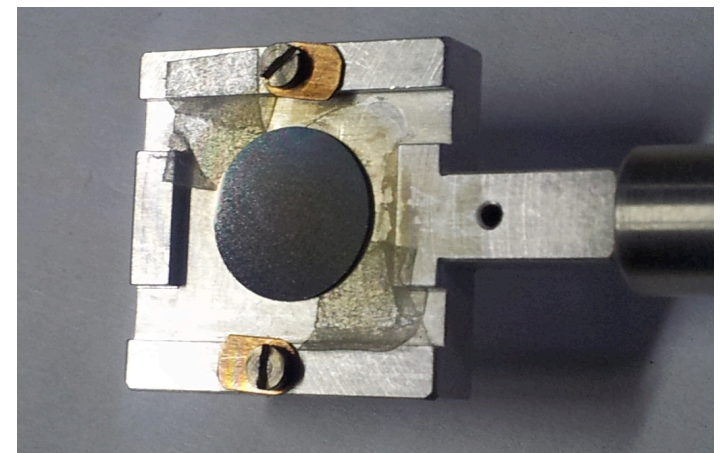
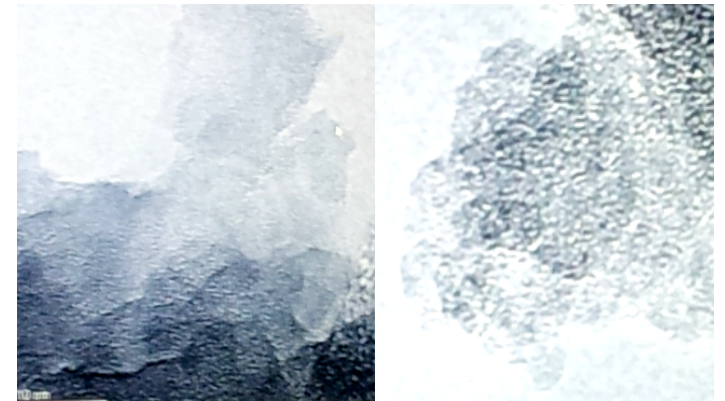
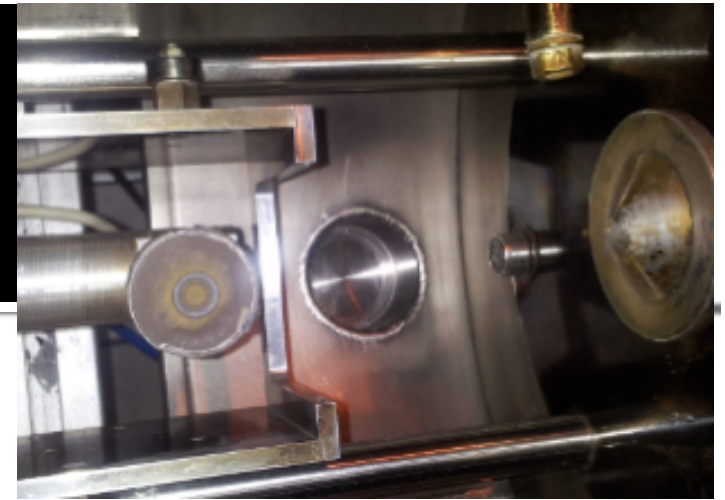
- We use an amorphous silicate $(\text{Fe}_x, \text{Mg}_{1-x})_2\text{SiO}_4$. A “rough” surface prepared from naturally occurring San Carlos Olivine.
- Chosen because it has deeper adsorption sites than crystalline silicates (Perets et al. 2005), which may enhance formation.
- $T_{\text{sample}} = 5 - 300 \text{ K}$.



Mg Fe silicate (Arizona, San Carlos Apache Reservation)

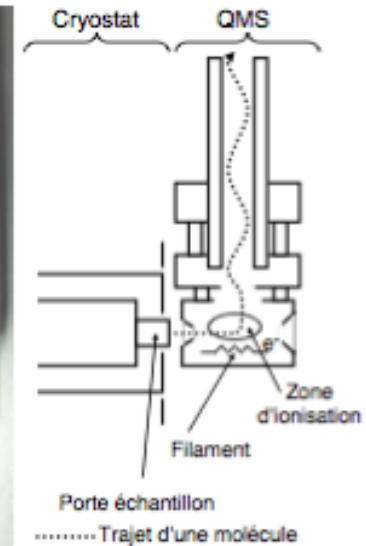
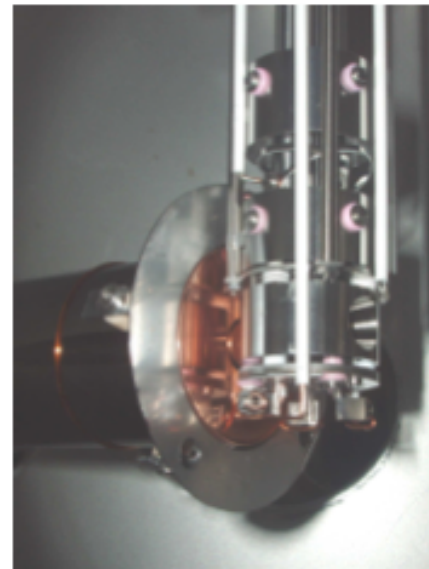
The sample

- Also, Forsterite (Mg_2SiO_4) and Fayallite (Fe_2SiO_4) samples have been prepared by laser ablation on Si disks in collaboration with the Jena astrophysical institute (Germany).
- We will have one amorphous and one crystalline sample of each type (annealed to $> 750\text{ C}$).
- Diagnostics: TEM, EDX, IR.
- The goal is to study the chemisorption effects due to surface morphology and composition on H_2 formation.



The detectors: QMS (Quadrupole mass spectrometer)

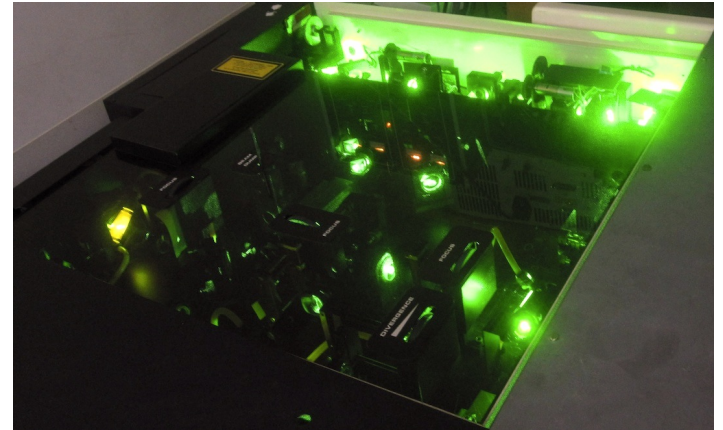
- This QMS can measure masses between 1 and 50 amu.
- It gives a good idea on the absolute sticking of atoms and molecules, but as it only distinguishes across masses it's not a reliable method to study formation.
- We use the King and Wells Method, in which the QMS measures the average density of particles in the gas phase in the main chamber.



Amiaud, 2006

The detectors: REMPI-TOF (Resonantly Enhanced Multiphoton Ionization)

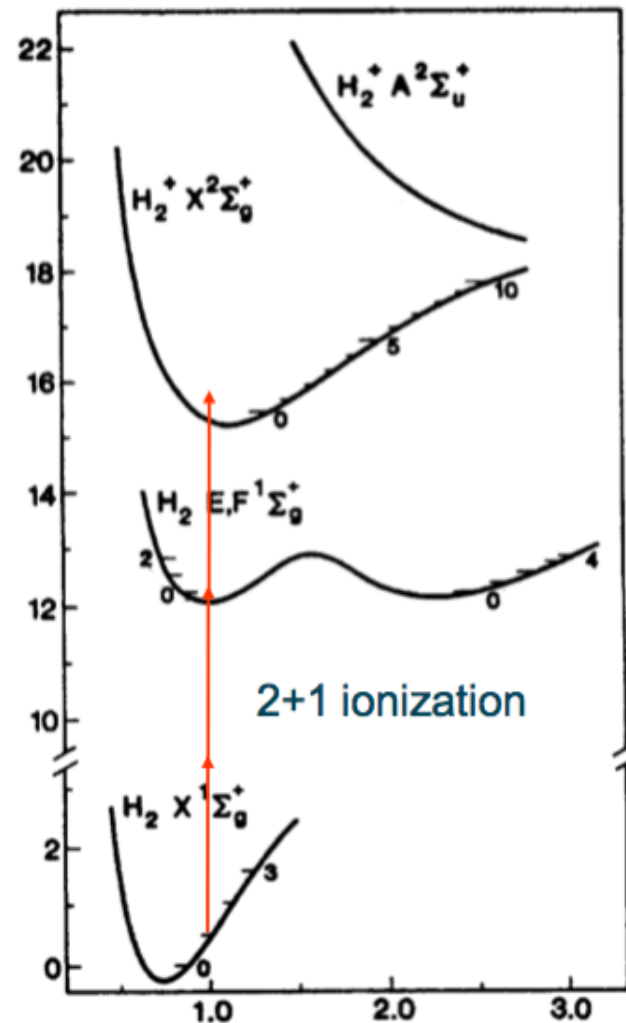
- REMPI is a quantum selective detector.
- REMPI (2+1) works with 3 photons at 220 nm. 2 are used to excite and one to ionize formed D_2 at $v''=4, J''=2$.
- 1 mJ pulsed laser (20 Hz).
- The laser is calibrated in an annex setup at higher D_2 pressure.



NdYAG laser @ Cergy

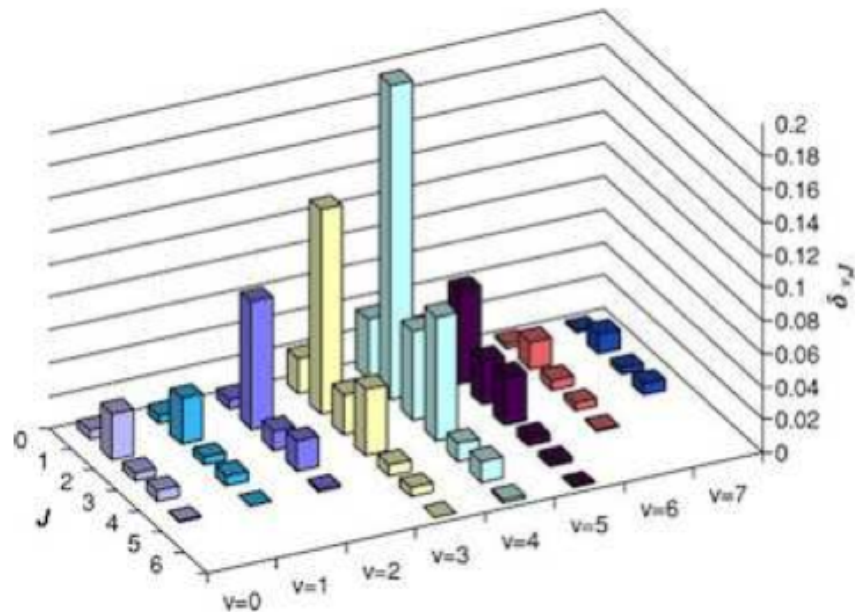
Internal excitation of nascent H_2

- REMPI (2+1): H_2^+ (E, F state)
- Beam is cooled down so that D atoms are in rovibrationally grounded, and only newly recombined D_2^* is detected.



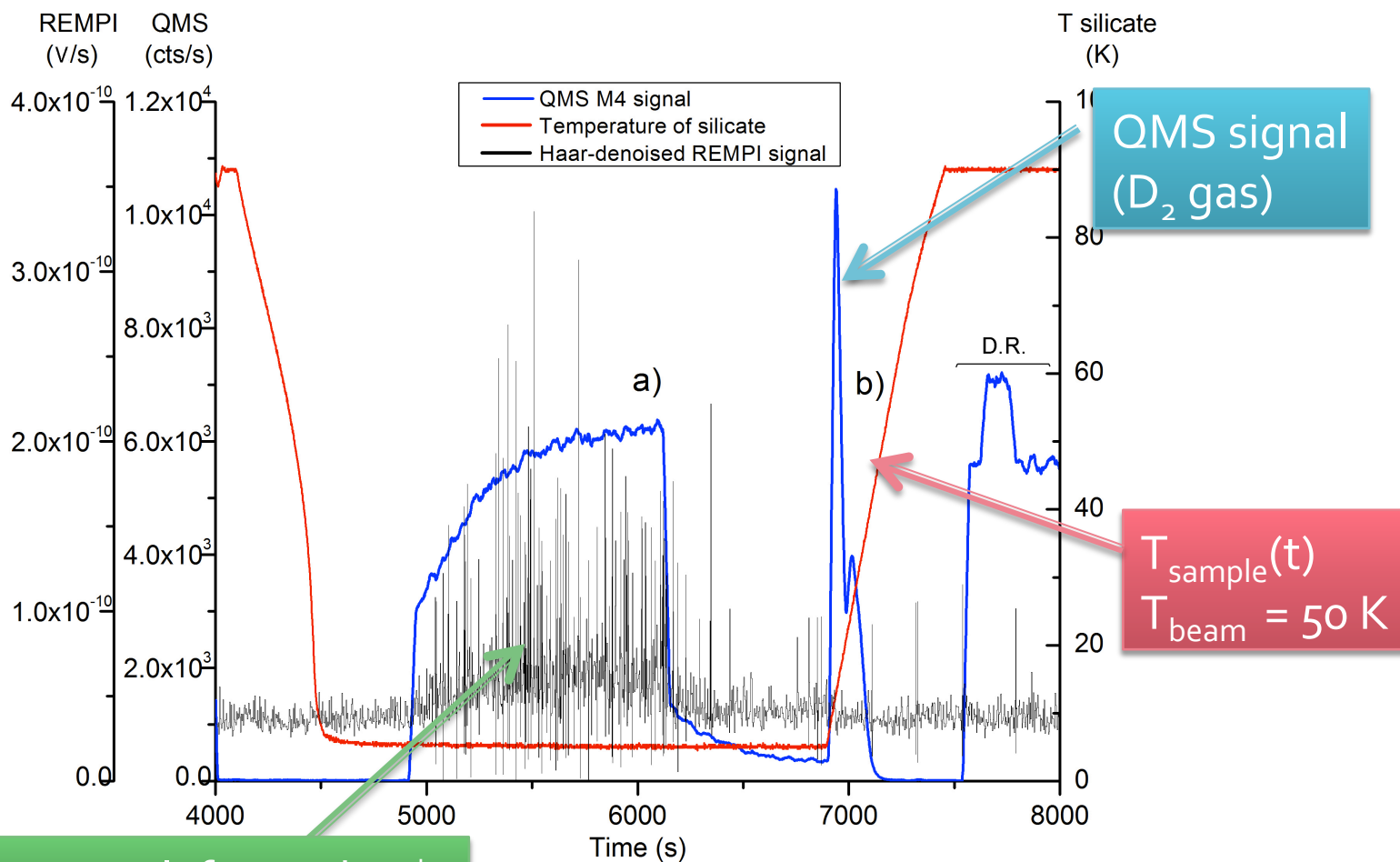
Internal excitation of nascent H_2

- For REMPI, the photon wavelength is tuned so that it ionizes only the $v''=4, J''=2$ level, predicted to be the most populated rovibrational level after formation.
- It's also a convenient state because of its Frank Condon factor.



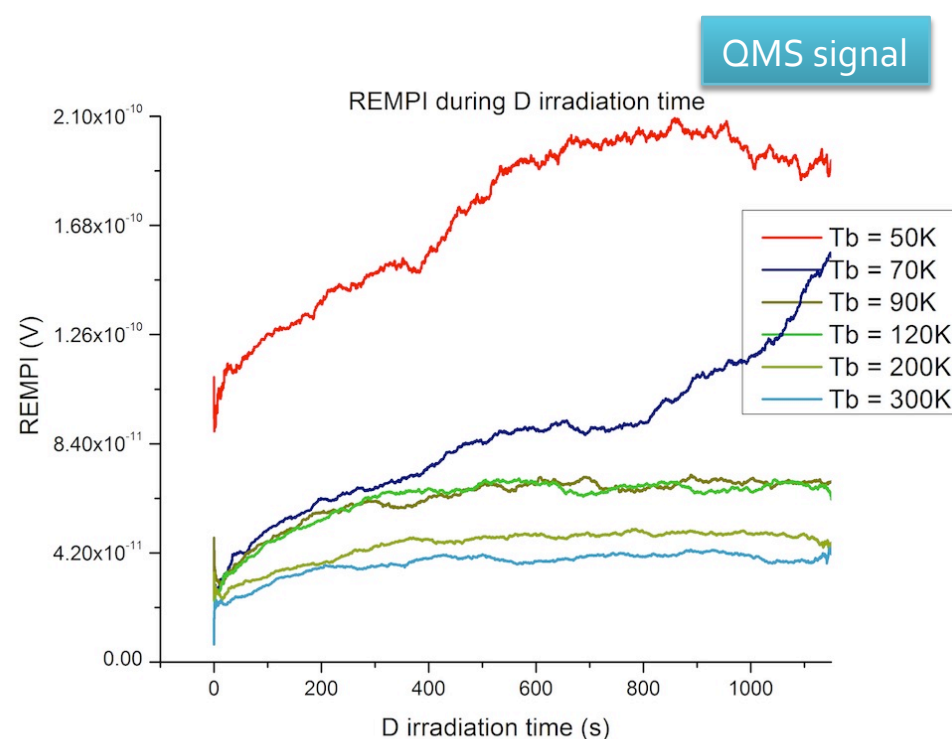
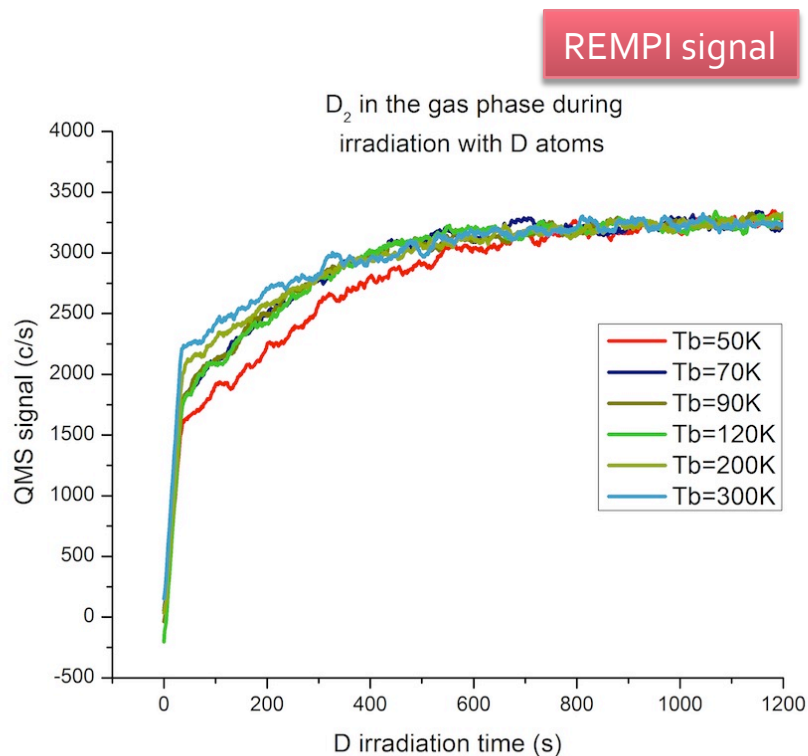
Rovibrational distribution of molecular hydrogen after formation extrapolated from HD at 15 K (Creighan 2006, Islam 2010).

Experiments: T_{beam} effect on formation



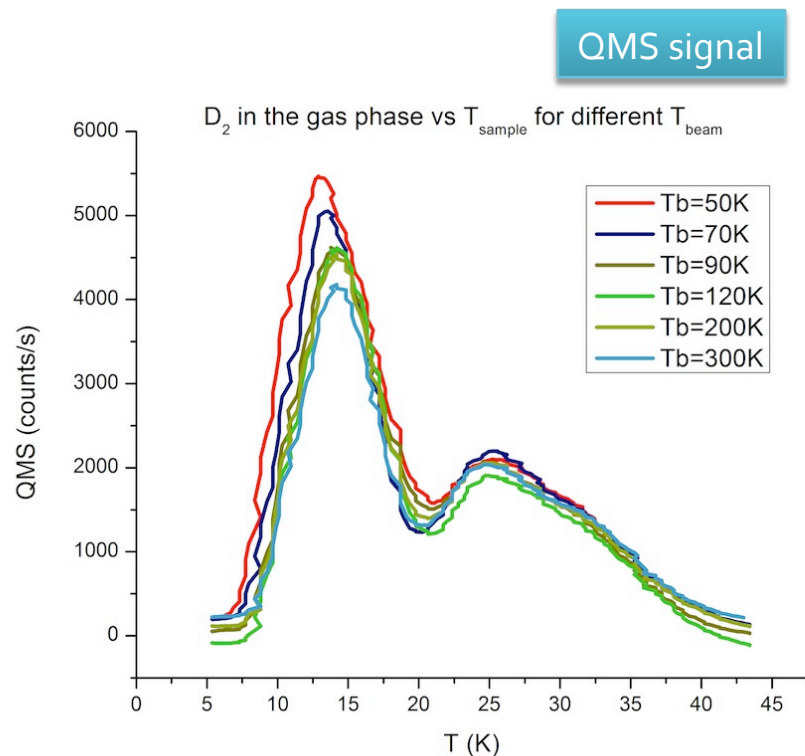
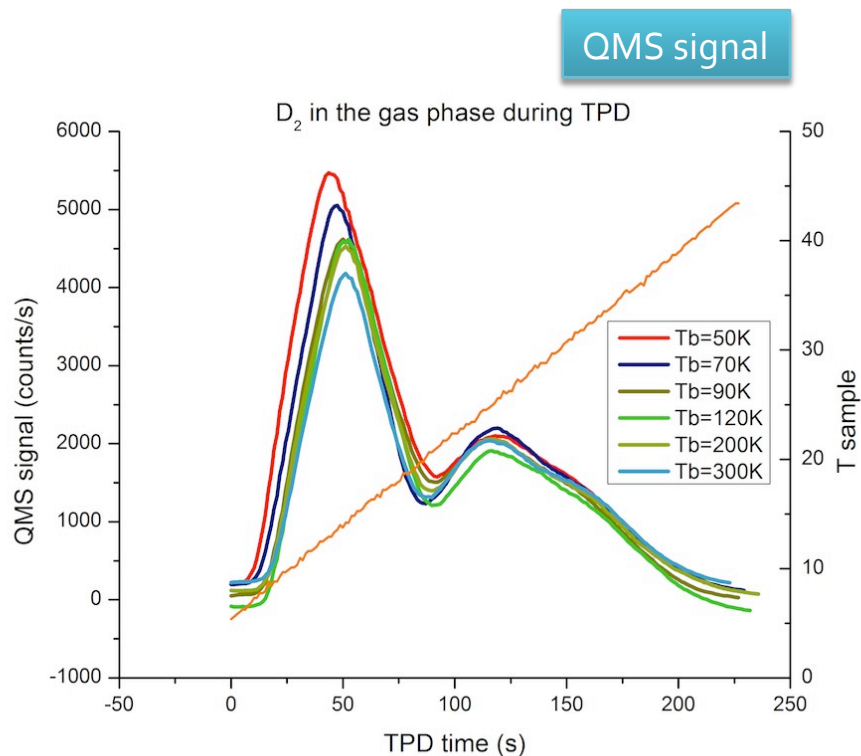
REMPI-TOF signal (formed D_2^*)

Experiments: T_{beam} effect on formation



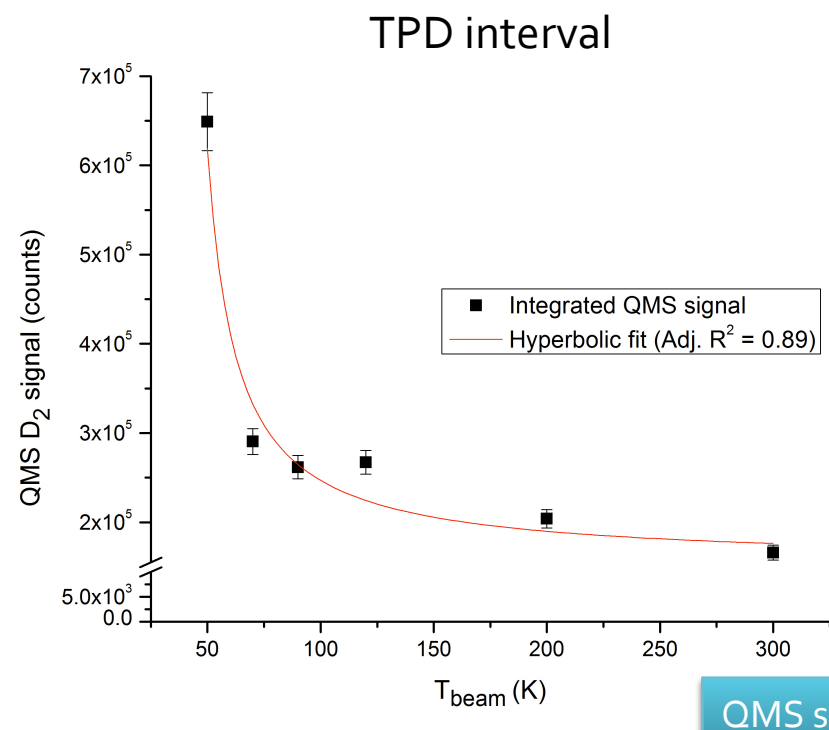
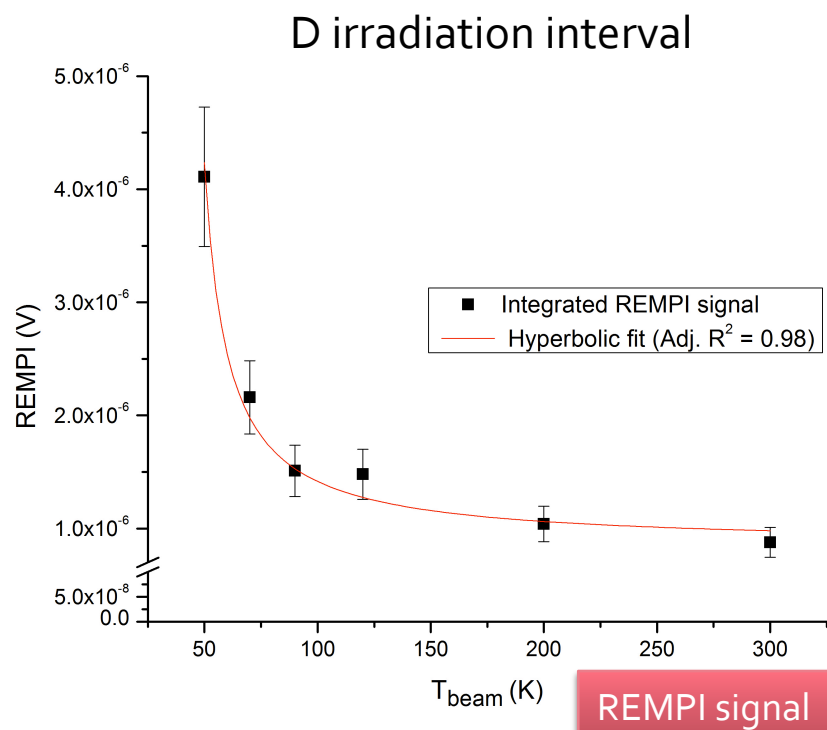
- QMS: largest D sticking from $T_b = 50\text{K}$, but slowest increase until recombination/sticking steady state. Growth curve may be dominated by flux of undissociated D_2 (greatest at lower T_b).
- REMPI: recombination during this time enhanced for $T_b = 50\text{K}$. Unlike the QMS, it discriminates the formed D_2 from undissociated D_2 .

Experiments: T_{beam} effect on formation



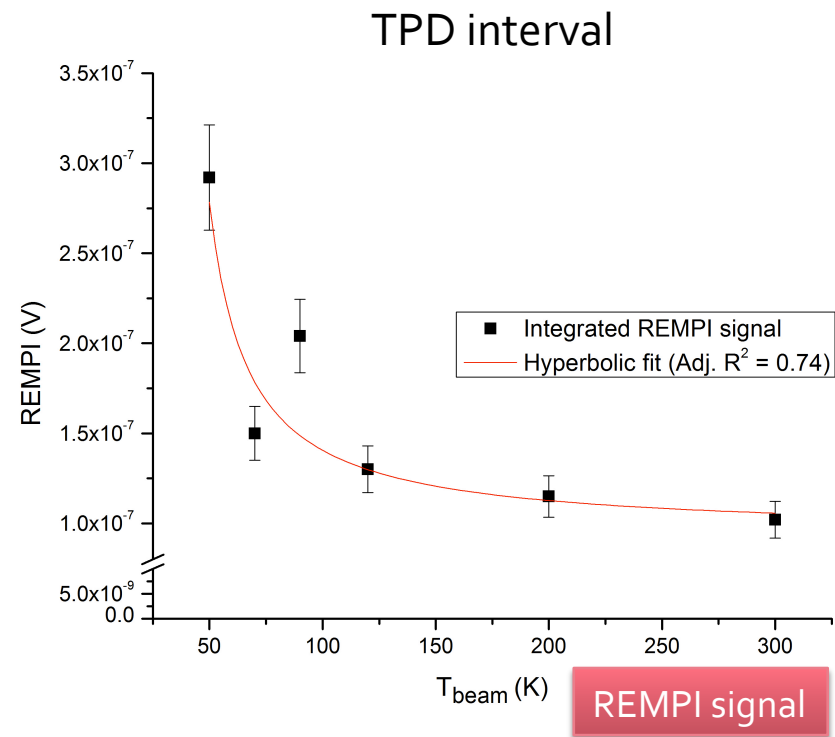
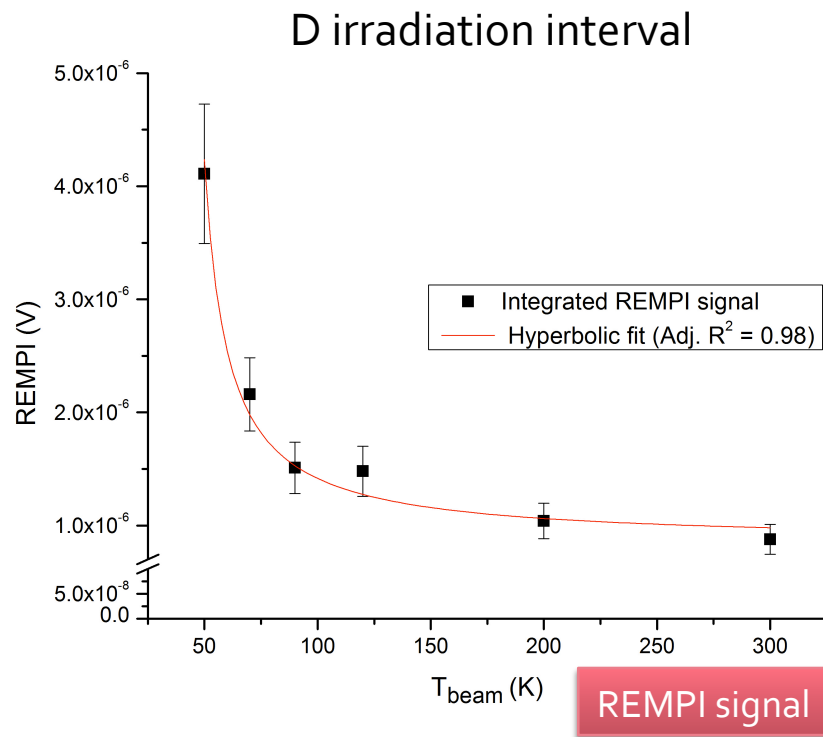
- QMS: TPD curves showing the largest sticking of D_2 (recombined and undissociated) was at $T_b = 50\text{K}$. TPD is desorbing the particles that were more strongly bounded to the surface, and it confirms what D atoms have the largest sticking on silicate for colder beam, i.e. $T_b = 50\text{K}$.
- QMS: TPD curves vs T_{sample} can be analyzed by desorption models.

Experiments: T_{beam} effect on formation



- REMPI area during D irradiation: Lower beam temperatures enhance formation.
- QMS area during TPD: Lower beam temperatures enhance sticking of D atoms on the surface.
- **We confirm the positive correlation between sticking and D_2 recombination for $T_{\text{sample}} = 5\text{K}$ and $T_{\text{beam}} = 50\text{-}300\text{K}$, enhanced at lower gas temperatures.**

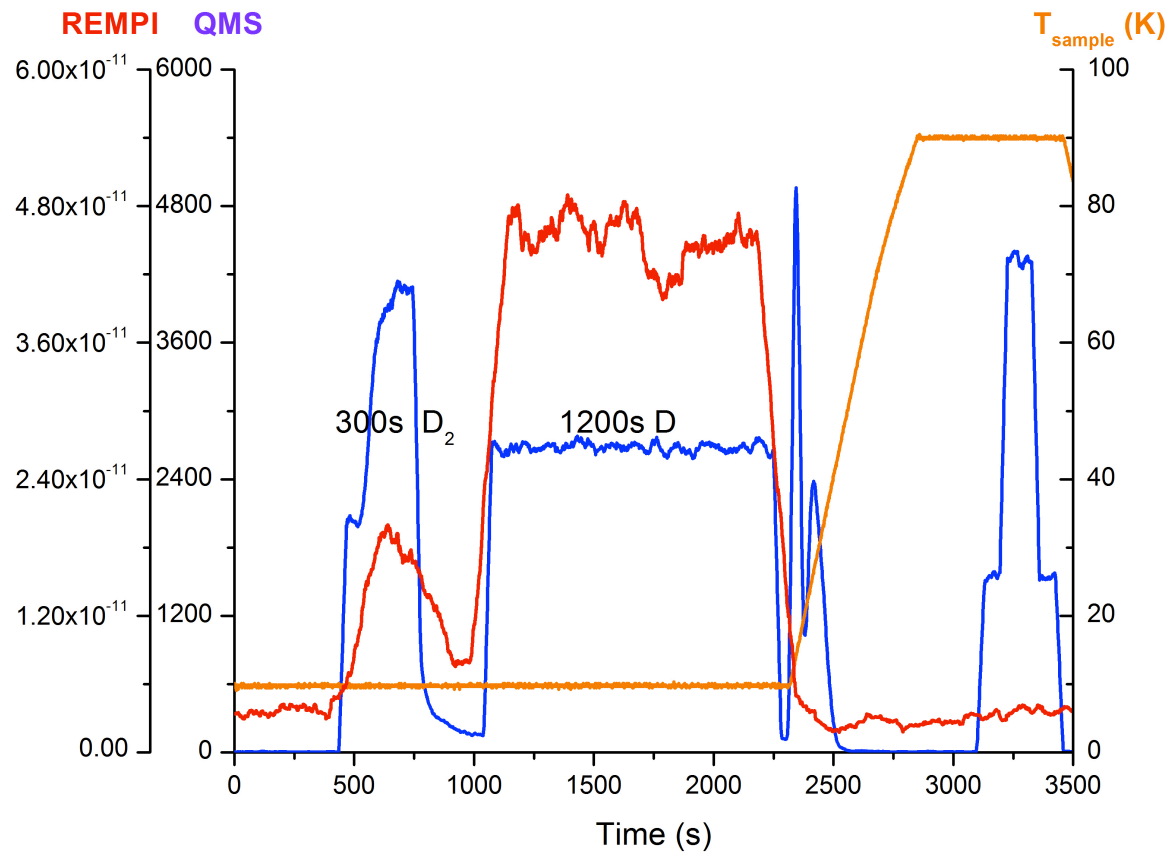
Experiments: T_{beam} effect on formation



- REMPI area during TPD: formation is activated by sample heating. The heating ramp helps surface D atoms overcome the diffusion barrier, allowing them to recombine
- Notice the REMPI signal is an order of magnitude weaker here than during D irradiation.

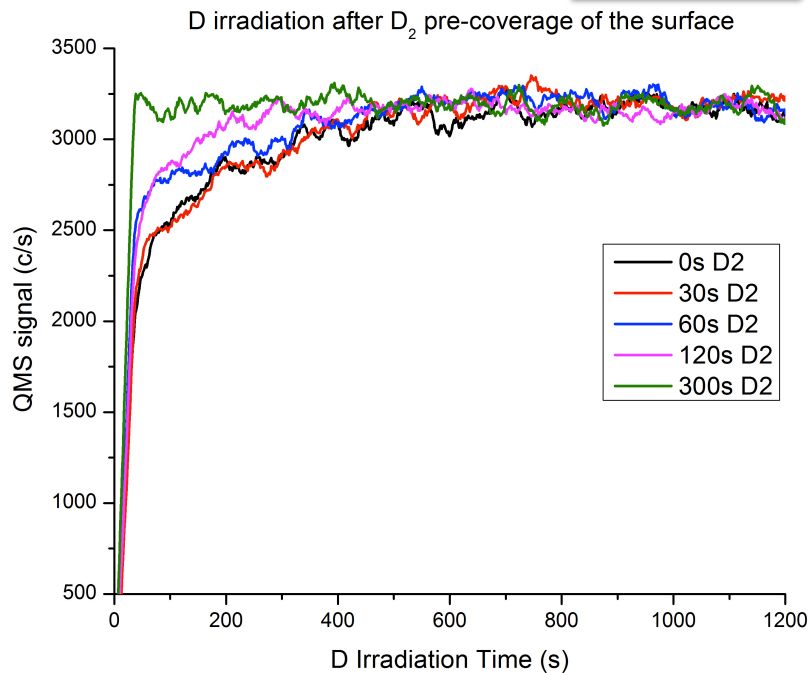
Experiments: D_2 pre-coverage effect on formation

- D_2 or H_2 on the surface can modify the process of accretion or diffusion (Govers et al., 1980; Tielens, 2005).

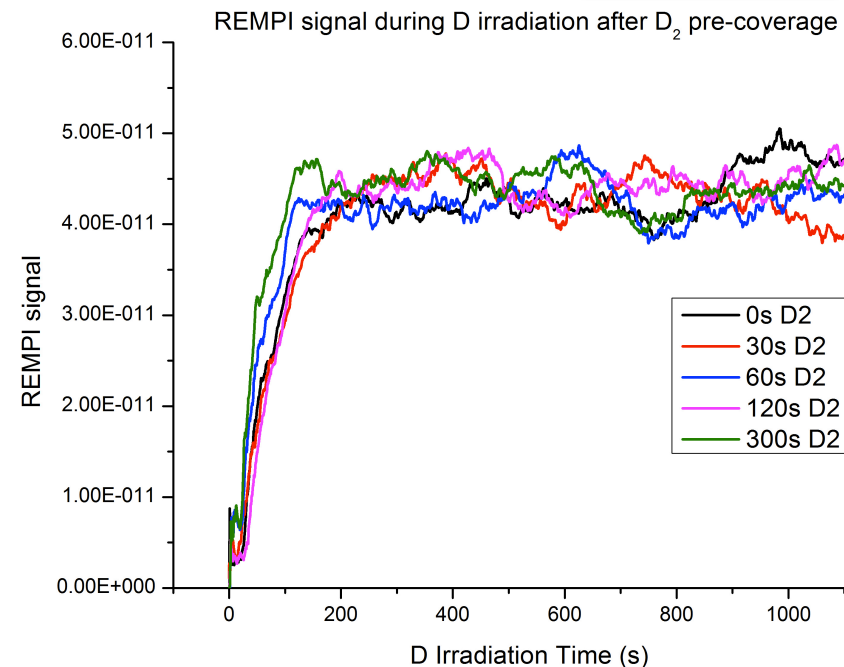


Experiments: D₂ pre-coverage effect on formation

QMS signal



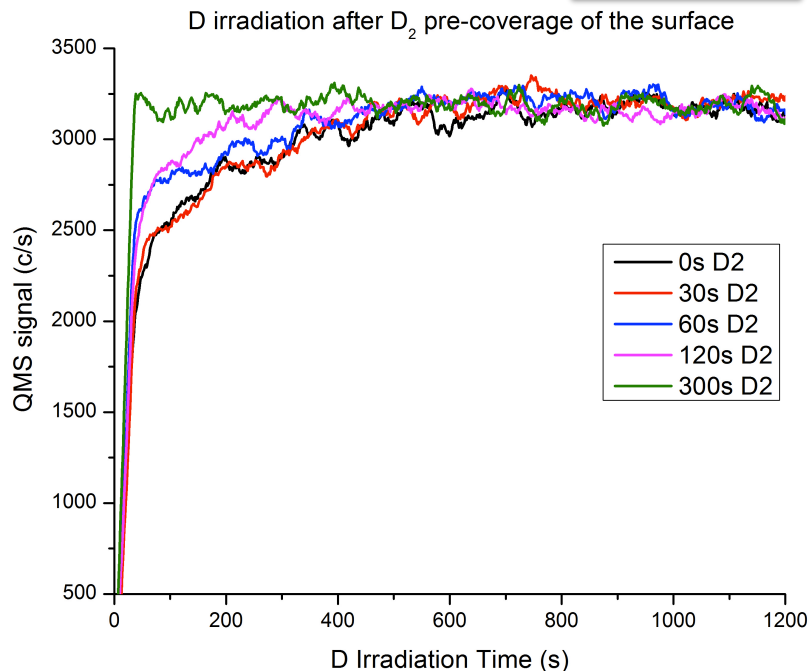
REMPI signal



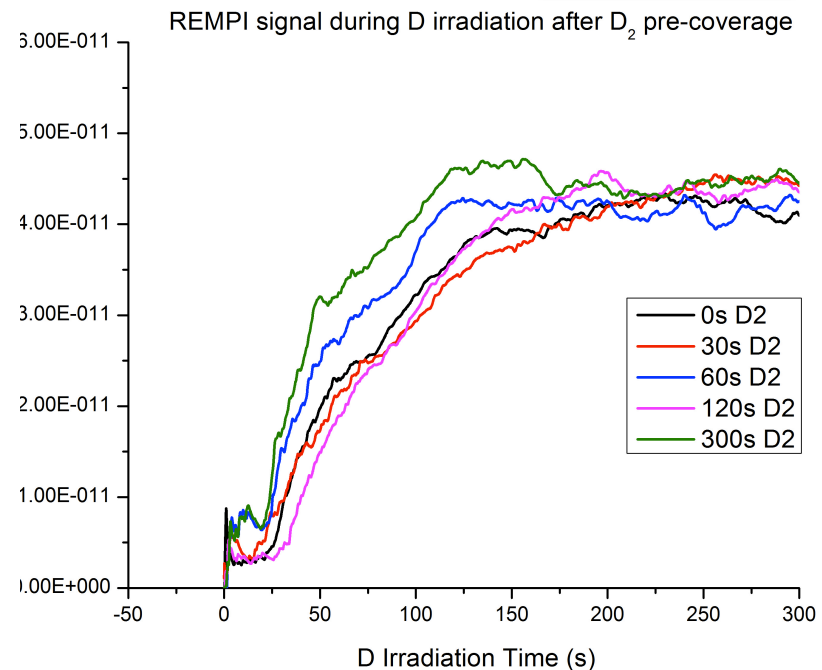
- QMS: growth curve depends on pre-coverage of D₂. Surface pre-coverage with 300s D₂ reaches a recombination steady state faster.
- REMPI: the REMPI signal confirms this being greatest for 300s of D₂ pre-coverage.

Experiments: D_2 pre-coverage effect on formation

QMS signal



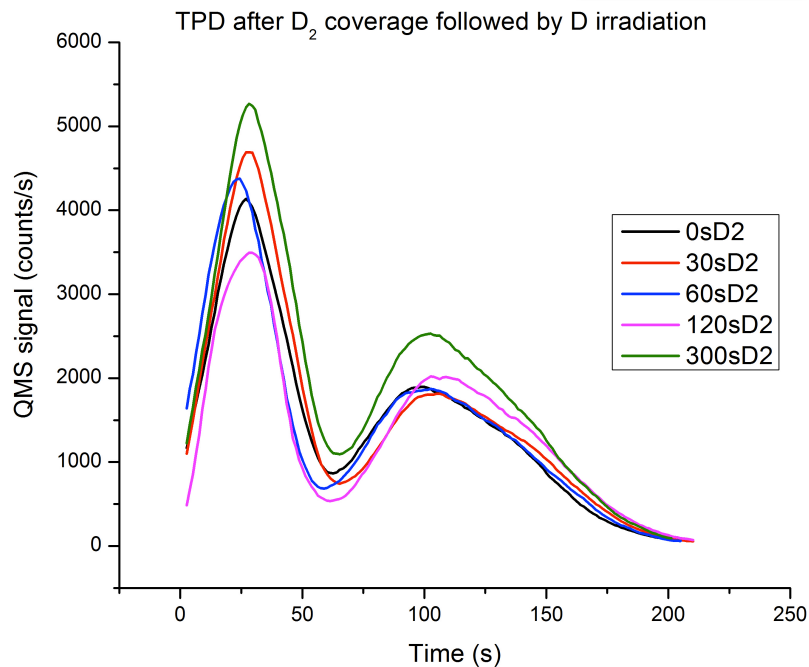
REMPI signal



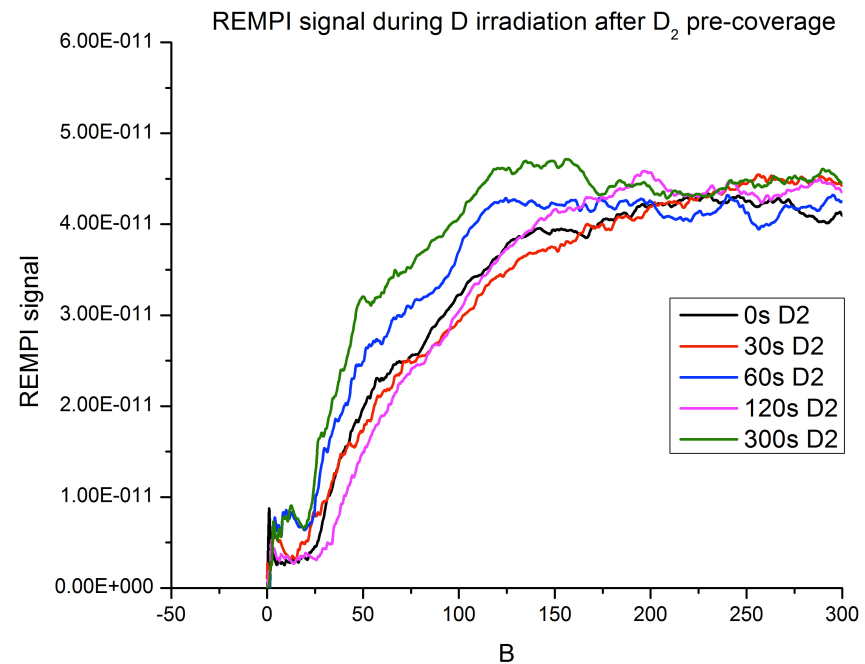
- REMPI: The effect of D_2 coverage seems to be most prominent during the first 170s of irradiation with D atoms.
- After this, the REMPI curves converge with a more homogenous coverage (both D and D_2).

Experiments: D_2 pre-coverage effect on formation

QMS signal

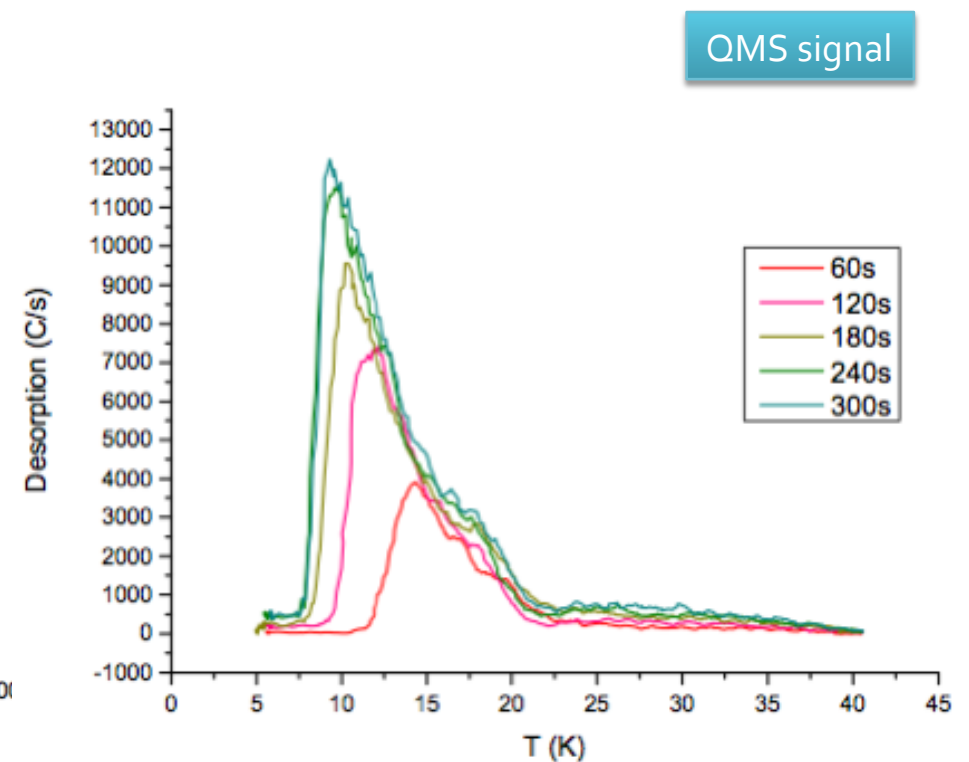
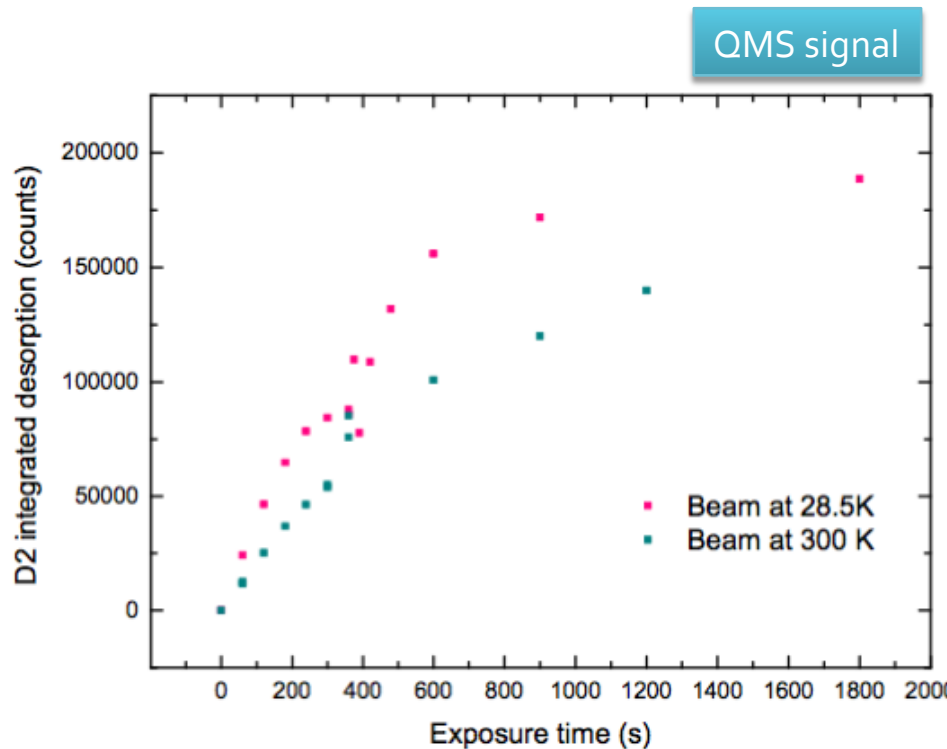


REMPI signal



- QMS: TPD curves show that coverage was non-linear with respect to irradiation dose on the surface. This is mostly affected by uneven waiting after D_2 dosage before D irradiation. D_2 is very volatile on the surface at 5K.
- REMPI: The effect of D_2 coverage on REMPI is correlated to the TPD areas.

Experiments: D_2 pre-coverage effect on formation



- Previous work on the sticking of D_2 molecules found that the evaporation flux at a surface temperature of 5.5K is not negligible.
- It is impossible to build a D_2 multilayer even at 5K.

Conclusions: formation mechanisms

- These new experiments probe $T_{\text{dust}} = 5\text{K}$ (fixed) and $T_{\text{beam}} = 50\text{-}300\text{K}$ ranges.
- We have focused on the effects of dynamical coverage of the surface on formation. We found that:
 - D_2 coverage enhances formation so D_2 formation is self-regulating.
 - Lower T_{beam} enhances formation as longer residence time enhances sticking.
- Hybrid mechanisms depend on the current coverage with D and D_2 on the surface.
 - When the grain is bare, at low coverage, few D atoms chemisorb on the surface. Initial irradiation causes a burst in formation for all T_{beam} . Eley-Rideal dominates.
 - Greater surface coverage with D and D_2 can lower barriers against diffusion, i.e. strong silicate binding sites are filled, so D atoms become more mobile and able to recombine. Langmuir-Hinshelwood dominates.

Conclusions: formation mechanisms

- Experimentally, the QMS is a detector that will give us only a partial view of the formation process.
- The QMS signal enlightens the absolute sticking and its evolution.
- However, since it only distinguishes across masses, and given the limitation in our dissociation source, it doesn't uncouple the signal D_2 due to recombination or from D_2 from the jet.
- REMPI-TOF has the great advantage of a quantum selective detector, profiting from the ro-vibrational excitation of newly formed D_2 on the dust.
- Its main disadvantages are the time-consuming wavelength tuning, variations of laser power from heating of the optics, and random drifts in the detector.